## Statistical Physics (PH312) HW #2, Fall 2018

Note that you SHOULD show the details of your work.

1. Consider a random walk in one dimension. In a single step the probability of a displacement between x and x + dx is given by

$$p(x)dx = \frac{1}{\sqrt{2\pi\sigma^2}} \exp\left(-\frac{(x-a)^2}{2\sigma^2}\right).$$

After *N* steps, the displacement of the walker is  $Y = x_1 + \cdots + x_N$  where  $x_i$  is the displacement in the *i*th step. Assume the steps are independent of one another.

- (a) What is the probability distribution for the displacement *Y*?
- (b) What are the average displacement,  $\langle Y \rangle$  and the variance,  $\langle (\Delta Y)^2 \rangle$ ?

(c) Repeat (a) and (b) for a random walk where the walker is equally likely to take a step with displacement in the interval  $d - a \le x \le d + a$  for a < d.

- 2. Consider a monoatomic ideal gas consisting of  $N_0$  molecules in a container of volume  $V_0$ . Show from the grand canonical ensemble picture that the number N of molecules contained in a region of volume V (anywhere inside the container) is given by the Poisson distribution, P(N, V).
  - (a) Show at first that  $\langle N \rangle = N_0 f$  and  $\langle N^2 \rangle \langle N \rangle^2 = N_0 f (1 f)$  where  $f = V/V_0$ .
  - (b) Show that if both  $N_0 f$  and  $N_0(1 f)$  are large numbers, P(N, V) assumes a Gaussian distribution.
  - (c) Further, if  $f \ll 1$  and  $N \ll N_0$ , show that

$$P(N,V) = \frac{\langle N \rangle^N}{N!} e^{-\langle N \rangle}$$

- 3. (a) Determine the grand partition function for a gas system of magnetic atoms that can have, in addition to the kinetic energy, a magnetic potential energy  $\mu_B H$  or  $-\mu_B H$  (with  $J = \frac{1}{2}$  and g = 2), depending on the orientation with respect to an applied magnetic field *H*.
  - (b) Derive the expression for the magnetization of the system,  $M_z$ .
  - (c) Calculate how much heat will be produced by the system when the magnetic field is reduced from H to zero at constant volume and constant temperature.
- 4. (a) For an isothermal-isobaric ensemble (NPT-ensemble), show that

$$P(V) = P(V^*) \exp\left(\frac{(V - \langle V \rangle)^2}{2kT(\partial V/\partial p)_{N,T}}\right).$$

where  $V^*$  is the most probable volume.

(b) For a grand canonical ensemble, show that

$$\langle (\Delta E)^2 \rangle = kT^2 C_V + \left(\frac{\partial U}{\partial N}\right)_{T,V}^2 \langle (\Delta N)^2 \rangle.$$

where  $\langle (\Delta E)^2 \rangle \equiv \langle E^2 \rangle - \langle E \rangle^2$  and  $\langle (\Delta N)^2 \rangle \equiv \langle N^2 \rangle - \langle N \rangle^2$ . (c) For a grand canonical ensemble, show that

$$\langle NE \rangle - \langle N \rangle \langle E \rangle = \left( \frac{\partial U}{\partial N} \right)_{T,V} \langle (\Delta N)^2 \rangle$$

and that

$$\langle (\Delta J)^2 \rangle = kT^2 C_V + \left\{ \left( \frac{\partial U}{\partial N} \right)_{T,V} - \mu \right\}^2 \langle (\Delta N)^2 \rangle$$

where  $J = E - N\mu$ .